



FSBSI «Marine Hydrophysical Institute of  
RAS»

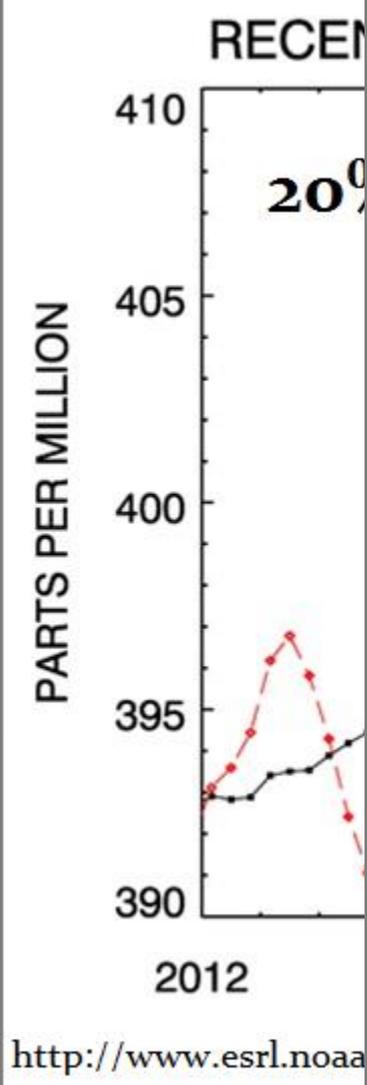
# CARBONATE SYSTEM TRANSFORMATION IN THE SEVASTOPOL BAY (THE BLACK SEA)

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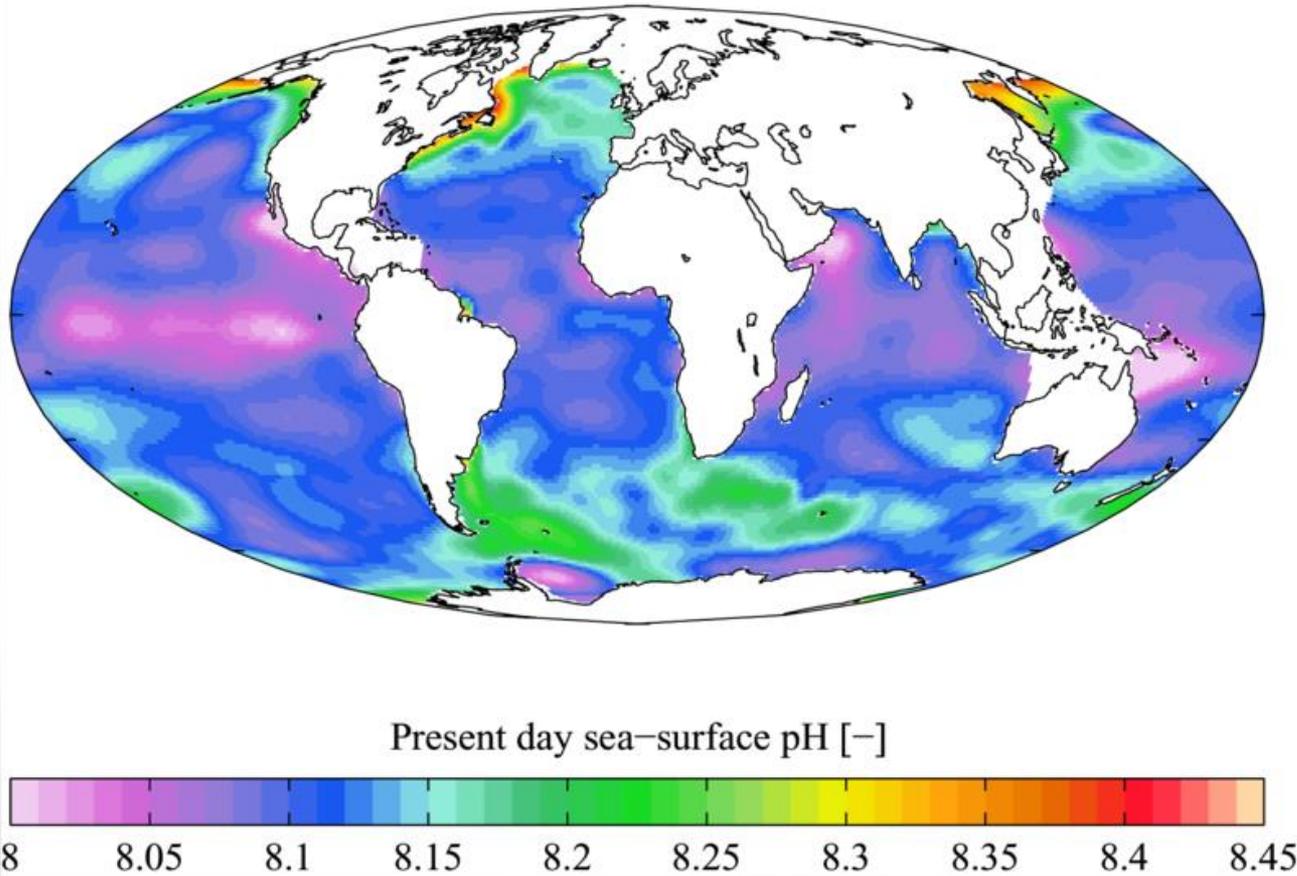
EMECS'11 SeaCoasts XXVI

Managing risks to coastal regions and communities in a changing world

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**pronounced carbon cycle transformation in coastal areas.**



**Acidification and CO<sub>2</sub> increase in the water column and appearance of oxygen minimum zones are reported for worldwide.**

**This makes ecological assessment of aquatic systems, including key cycles of elements, an important social and scientific task.**

# Main line

to present information on dynamics of the inorganic part of the carbon cycle and its transformation in the Sevastopol Bay from 1998 – 2015:

- ▣ long-term observations of the carbonate system's components;
- ▣ analysis of seasonal changes in the carbonate system's components;
- ▣ factors and processes influenced variations in the carbonate system's components;
- ▣ Sea-atmosphere CO<sub>2</sub> Fluxes (directions, changes)

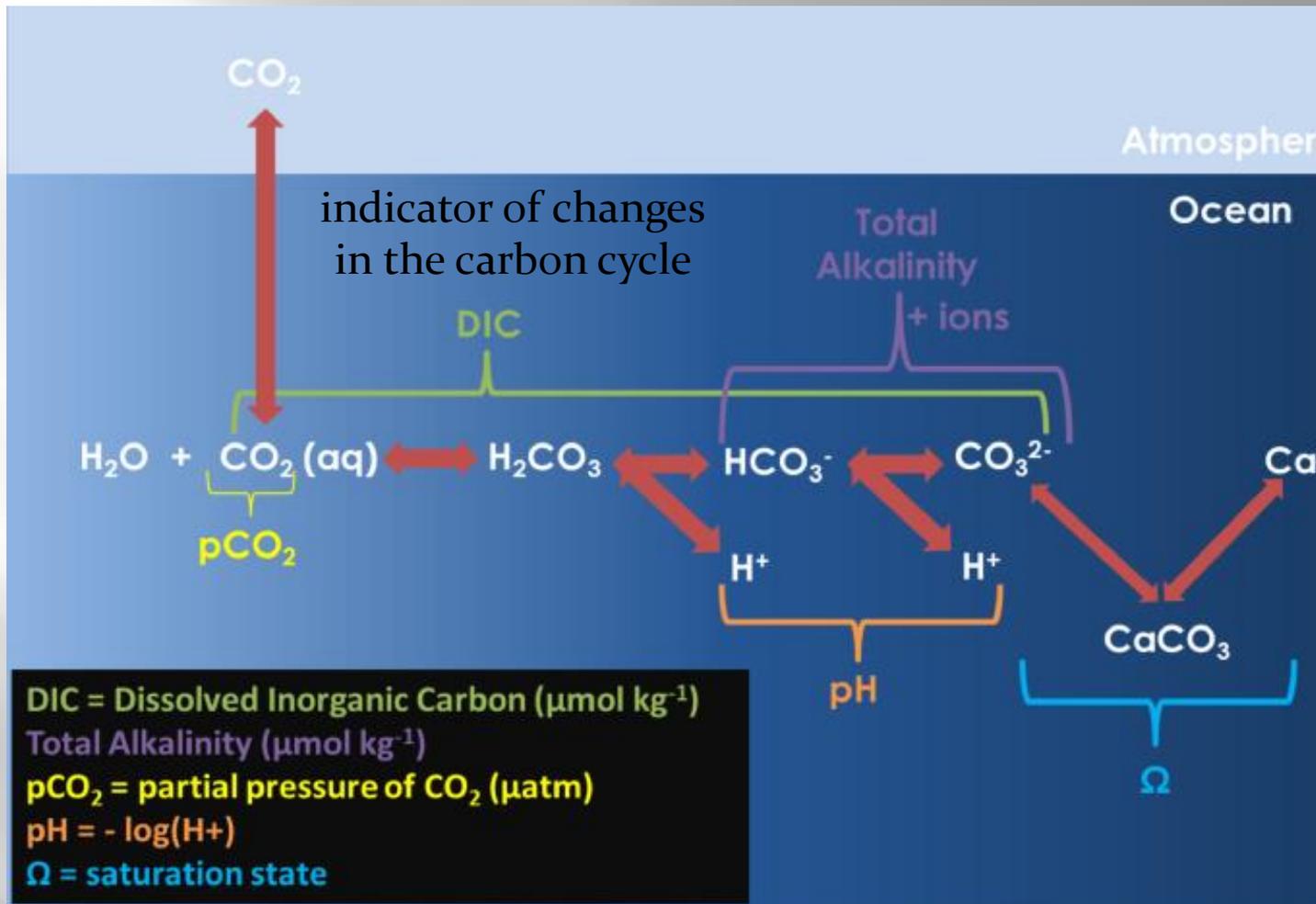
# Sevastopol Bay - semi-enclosed estuarine type bay.



The bay is under heavy anthropogenic pressure

It results in negative consequences for the bay's ecosystem

# The Carbon cycle in marine systems



<https://wattsupwiththat.com/2015/02/17/claim-versus-pr-hype-doesnt-hold-up-satellite-images-reveal-ocean-acidification-from-space/>

# Main processes result in variability of components of the carbon cycle

(1) photosynthesis:

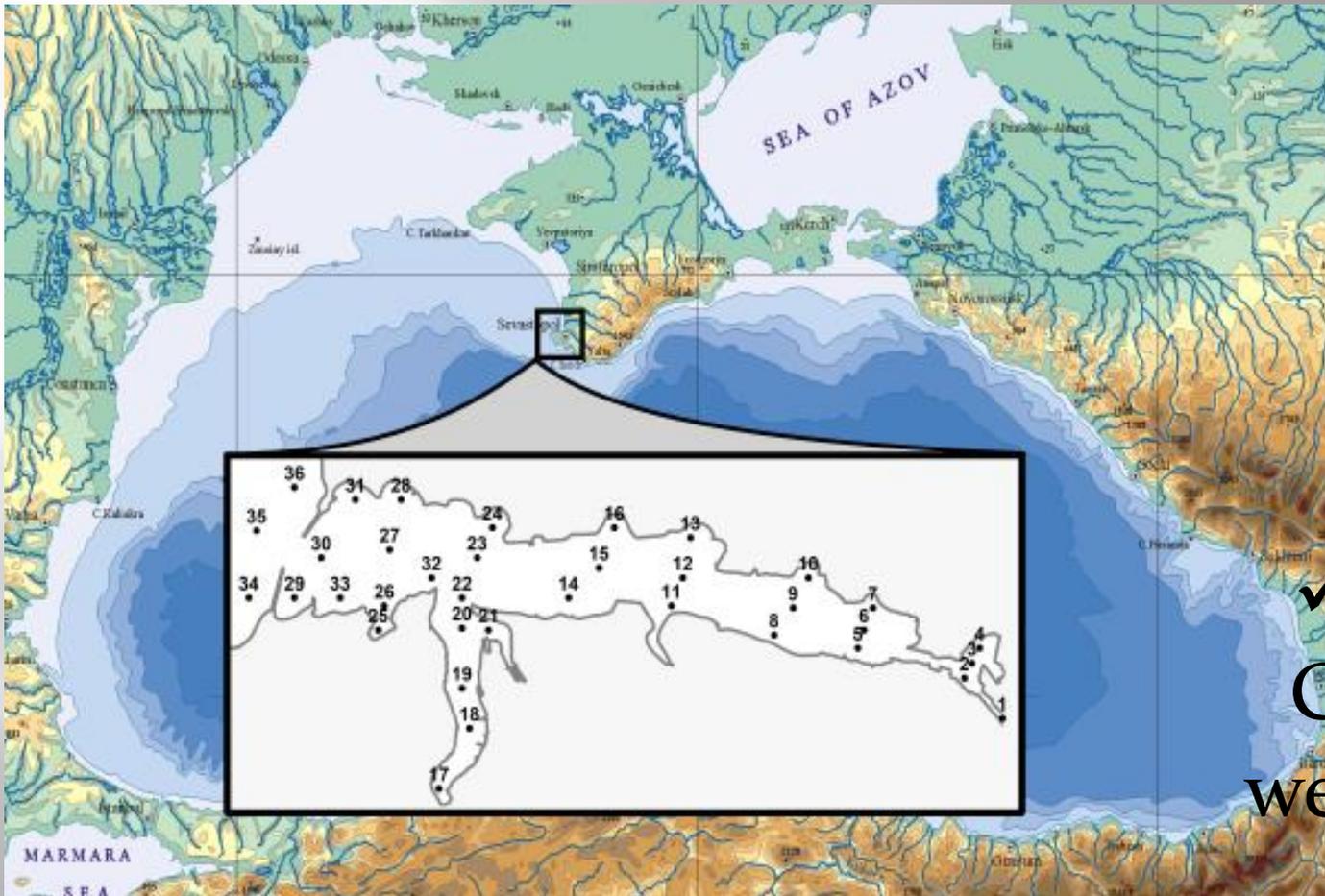


(2) biogeochemical organic carbon oxidation:



(3) physical process of  $\text{CO}_2$  transport

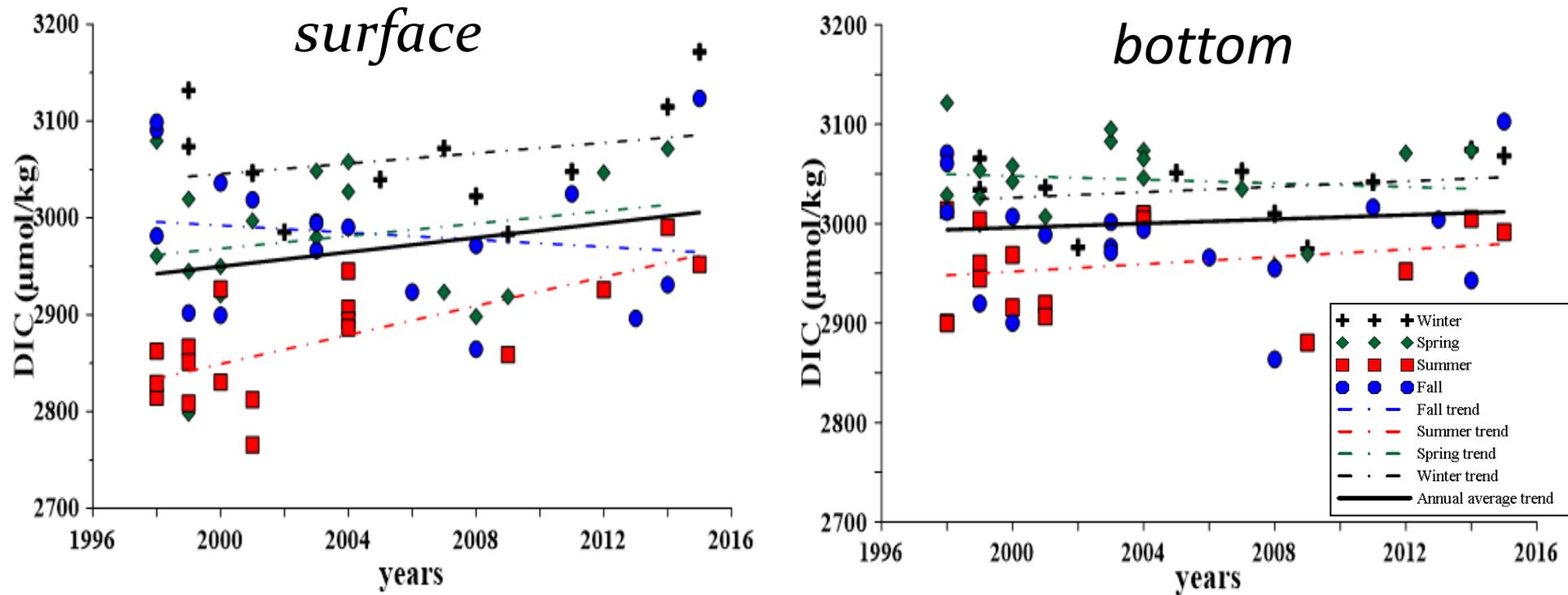
# *Sampling in the Bay of Sevastopol*



Determined parameters:  
✓ pH, alk. – analytically;

✓  $\text{CO}_2$ ,  $\text{HCO}_3^-$ ,  $\text{CO}_3^{2-}$ ,  $\text{pCO}_2$  – were calculated

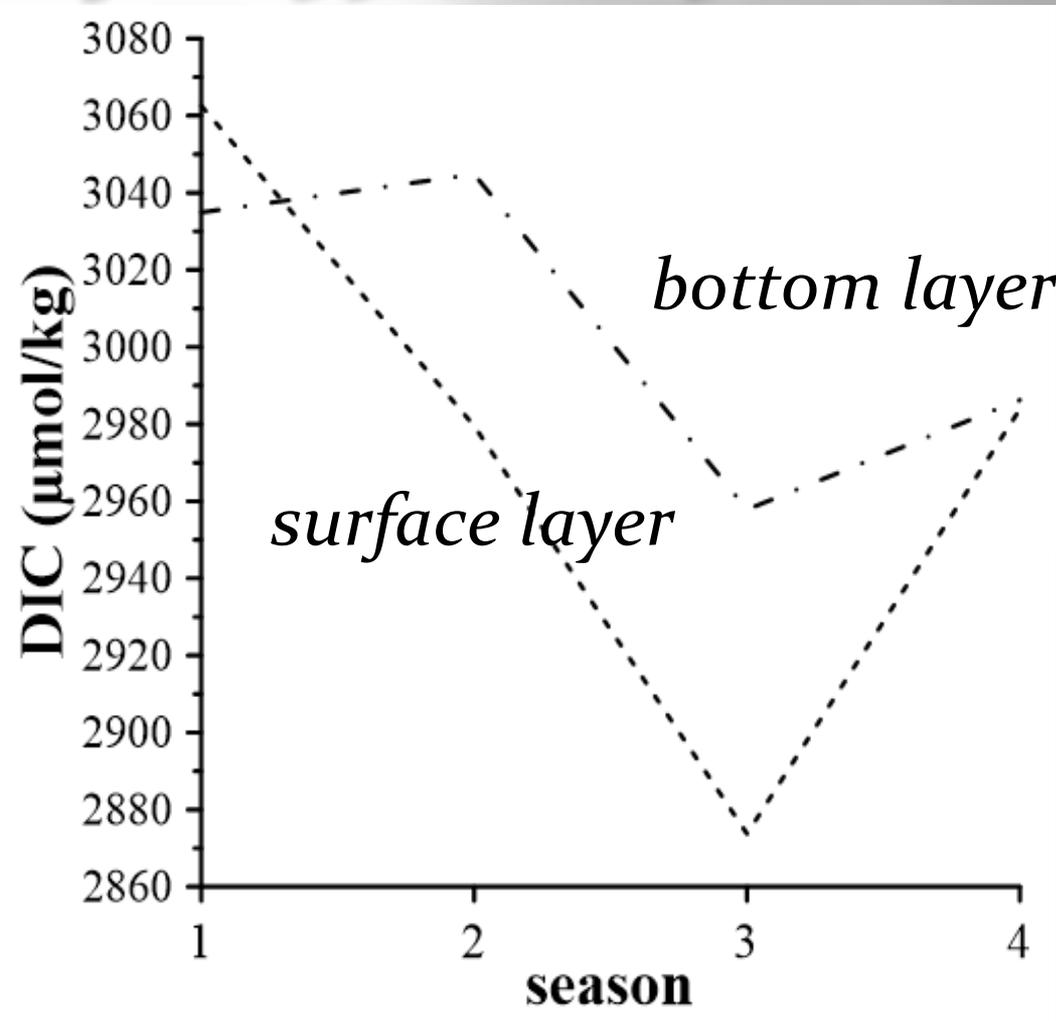
# Variations in the DIC in 1998 – 2015



DIC changes are insignificant and reveal up to 2% increase from 1998 – 2015.

This is evidence for the bay's carbonate system recovers.

# Seasonal DIC variability averaged for 1998 – 2015



1 – Winter, 2 – Spring,  
3 – Summer, 4 – Fall

**Winter:** DIC surface is maximum (ventilation,  $\uparrow\text{CO}_2$  solubility).

**Spring:** DIC bottom is maximum; DIC surface – high values due to seasonal flood of r. Chernaya (riched with Corg,  $\text{CO}_2$  and PSM)

**Summer:** DIC is minimal (photosynthesis,  $\downarrow\text{CO}_2$  solubility, strong vertical stratification).

**Fall:**  $\uparrow$  DIC, DIC bottom = DIC surface (vertical ventilation, Corg oxidation in the bottom)

# Variations in other components of the carbonate system

Species	Surface Layer	Bottom Layer	
$\text{HCO}_3^-$ ,	up to 4%	up to 1%	} const
$\text{CO}_2$	up to 20%	down to 2%	
$\text{CO}_3^{2-}$	down to 17%	up to 2%	
pH	down to 1%	down to 0,1%	

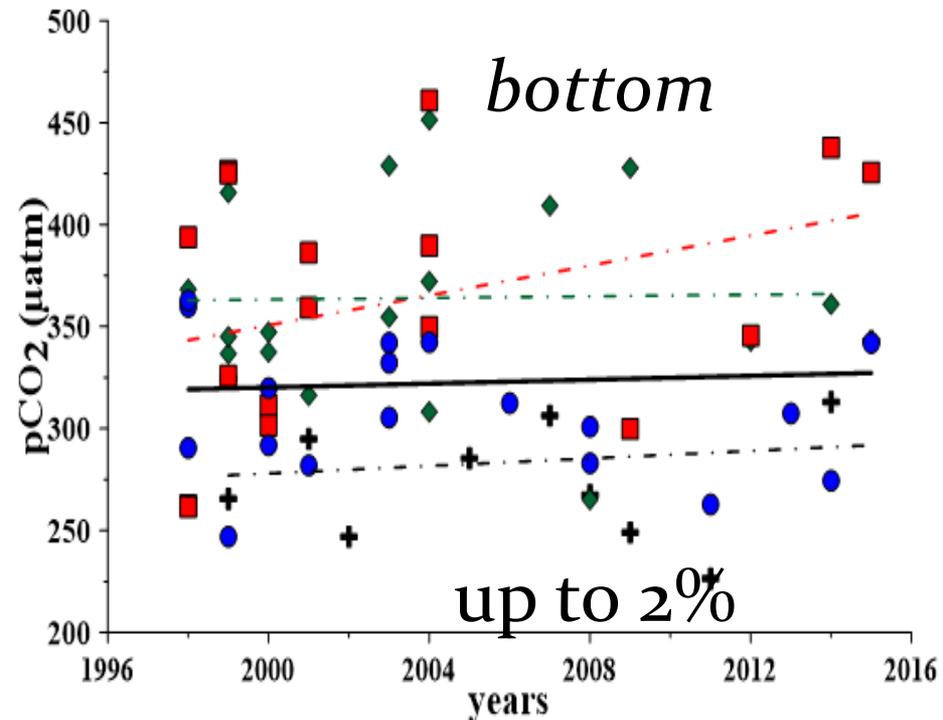
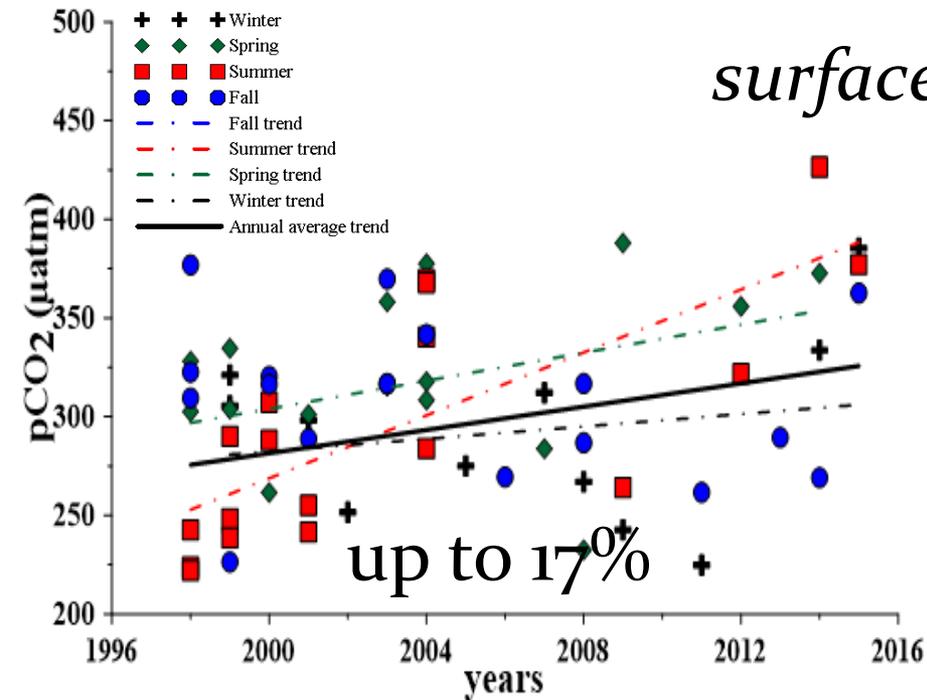


Additional flux  
of  $\text{C}_{\text{org}}$  from  
sewage and coast



Analytically negligible  
or ?  
Source of carbonates

# Variation in pCO<sub>2</sub>

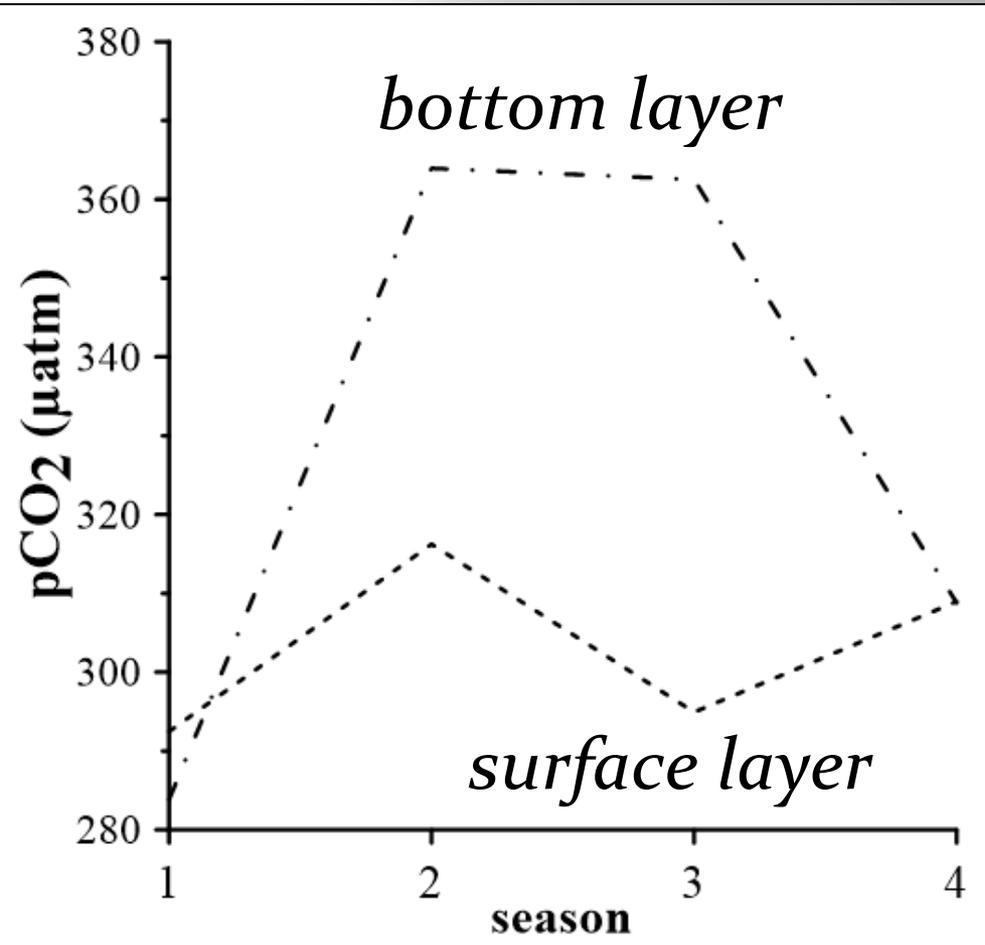


pCO<sub>2</sub> bottom > pCO<sub>2</sub> surface

**Surface:** a persistent upward trend in winter (by 9%), spring (by 19%) and especially in summer (up to 46%), due to ↑C<sub>org</sub> oxidation rate

**Bottom:** uprising trend is traced for summer (by 17%). In winter and spring, this upward trends are ~6 and 1% respectively. In fall, a 6% decrease of the pCO<sub>2</sub> is observed.

# Seasonal variations of $p\text{CO}_2$



1 – Winter, 2 – Spring,  
3 – Summer, 4 – Fall

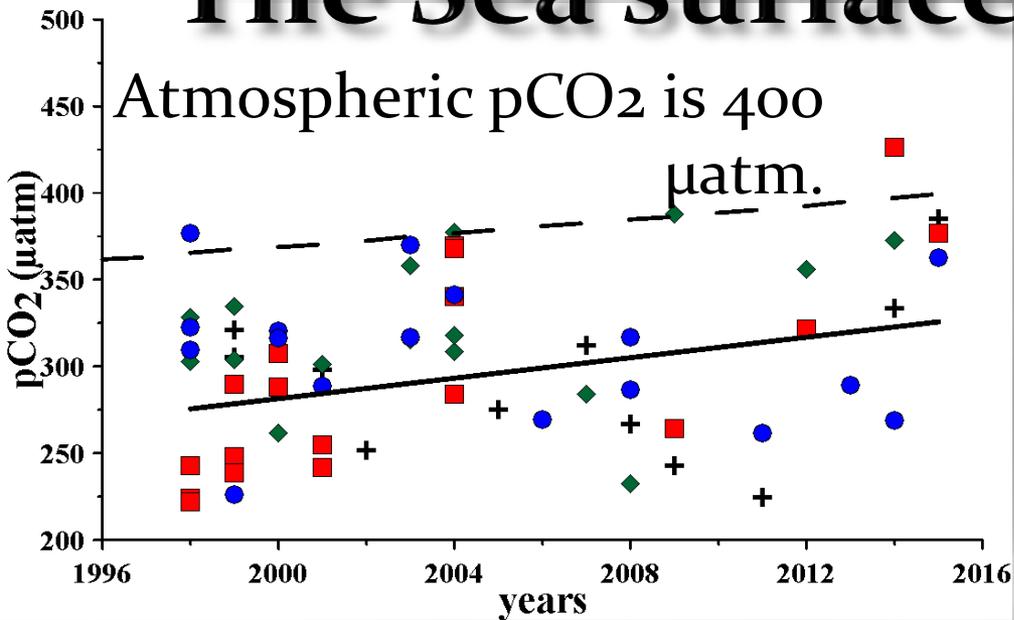
**Winter:**  $p\text{CO}_2$  minimum (vertical mixing,  $\downarrow$  Corg oxidation,  $\uparrow$  gas solubility and absorption of  $\text{CO}_2$  from the atmosphere).

**Spring:**  $p\text{CO}_2$  surface <  $p\text{CO}_2$  bottom.  $p\text{CO}_2$  is maximum in the surface due to influx of  $\text{CO}_2$  with r. Chernaya waters.

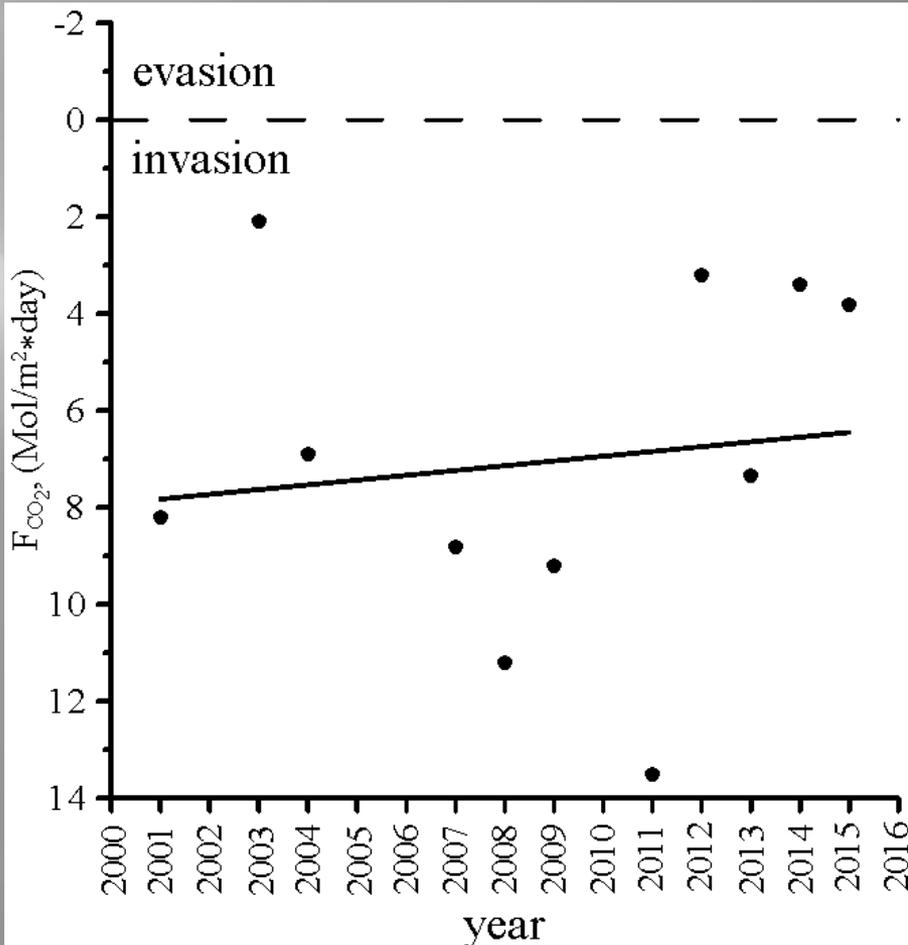
**Summer:**  $p\text{CO}_2$  surface  $\downarrow$  (eutrophication).  $p\text{CO}_2$  bottom is equal spring value (Corg oxidation). Maximal difference between layers (strong vertical stratification)

**Fall:**  $p\text{CO}_2$  surface =  $p\text{CO}_2$  bottom ( $\uparrow$  solubility of  $\text{CO}_2$  in cooling waters at surface; vertical mixing).

# The Sea surface Flux of pCO<sub>2</sub>



Bay's waters mainly absorb CO<sub>2</sub> from the atmosphere (that is unusual for the coastal ocean



Invasion (waters absorb CO<sub>2</sub> from the atmosphere) is typical, but this ability to consume CO<sub>2</sub> from the atmosphere has decreased by 20% from 2001 to 2015.

# Summary

- ◆ DIC changes are analytically insignificant. Increase of  $p\text{CO}_2$ ,  $\text{CO}_2$  and  $\text{HCO}_3^-$  in the surface indicates anthropogenic pressure.
- ◆ In warm seasons the main processes are Corg respiration in bottom and photosynthesis in the surface layers.
- ◆ In cold seasons, water dynamics and mixing promote system's recovery.
- ◆ Seasonal oscillations are far more pronounced, and reveal extremes for appearance of oxygen minimum zones.
- ◆ The surface waters still absorb  $\text{CO}_2$  from the atmosphere, but this ability has decreased by 20% from 2001 to 2015.

*Ecosystem Sevastopol Bay remain reversible, and the carbonate system can be restored.*

# THANKS FOR YOUR ATTENTION

This work has been funded from the RFBR research project N°16-35-60006 mol\_a\_dk "Long-term changes in the carbon cycle characteristics of the Sevastopol bay."

The total alkalinity was determined by direct titration by hydrochloric acid with a potentiometric end-point detection, pH was determined with a potentiometric pH-meter with NBS buffer solutions [1]. The standard deviation of measurement of 10 parallel samples did not exceed 0.02 pH units. Whereas  $\text{CO}_2$ ,  $\text{HCO}_3^-$ ,  $\text{CO}_3^{2-}$  concentrations and  $\text{pCO}_2$  values were calculated as suggested in [7]. The dissociation constants of carbonic acid, recommended by the Department of Marine Sciences of UNESCO, were used [7]. Boron was assumed a conservative part of salinity; therefore, the boron content was calculated from the salinity [4]. The effect of dissociation of water, phosphoric acid, sulfuric acid, hydrofluoric acid and other acids were not taken into account.